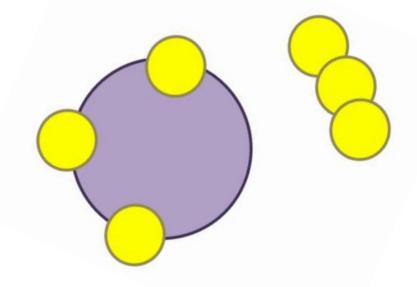
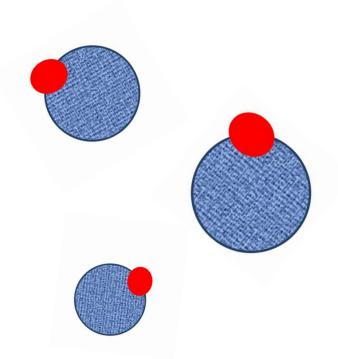


### Our Learning Objective

Develop models to
 describe the atomic
 composition of simple
 molecules and extended
 structures.



#### I can...

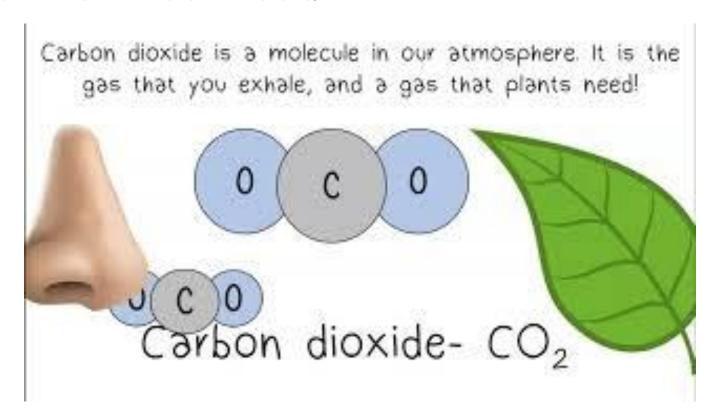


Identify that all matter is made out of atoms

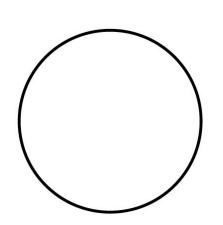
Identify that atoms combine to form molecules

 Compare and contrast molecules, compounds, elements, and mixtures

#### Review from Last Week:



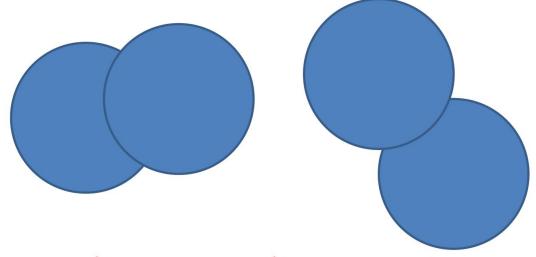
# An atom is the smallest unit of matter.



This is represented by a circle.

# Counting the number of atoms is easy!

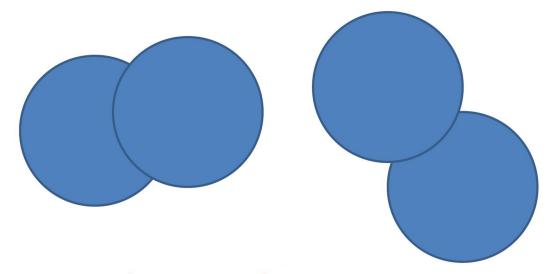
Just count how many circles!



How many atoms are in this example?

# Counting the number of atoms is easy!

Just count how many circles!



There are 4 atoms in this example

Each element has an atom that is completely unique.

How do we determine what element an atom is? (Hint: it's a subatomic particle.)

# Each element has an atom that is completely unique.

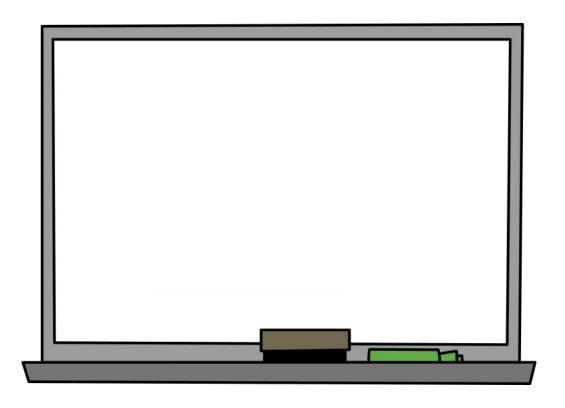
How do we determine what element an atom is? (Hint: it's a subatomic particle.)

The number of protons!



There are two different elements in this example. How many atoms are there?

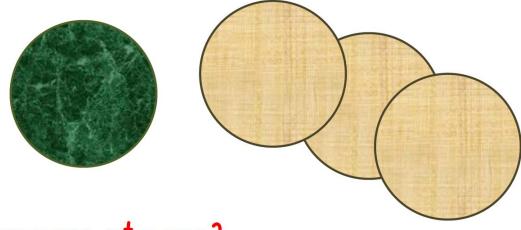
### Write the answer on your whiteboard!





There are two different elements in this example. How many atoms are there?

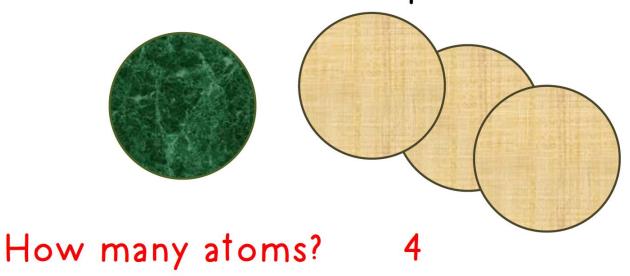
Count the number of atoms and elements in this example:



How many atoms?

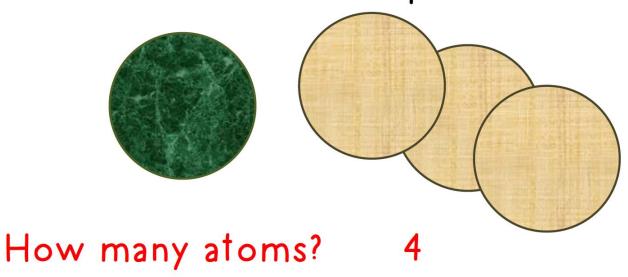
How many elements?

Count the number of atoms and elements in this example:



How many elements?

Count the number of atoms and elements in this example:



How many elements? 2

If you recall, a molecule is a group of 2 or more atoms. Any atoms!



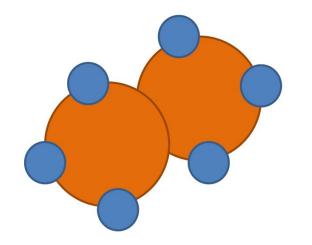
Is this a molecule?

If you recall, a molecule is a group of 2 or more atoms. Any atoms!



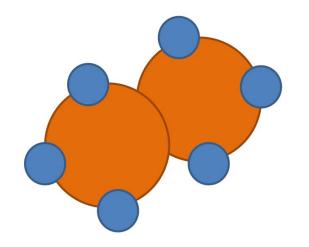
Is this a molecule?

No, because it is only one atom.



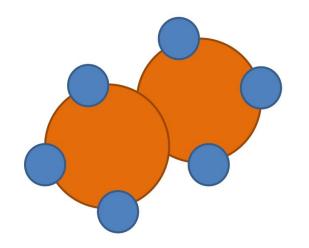
How many atoms?

How many elements?



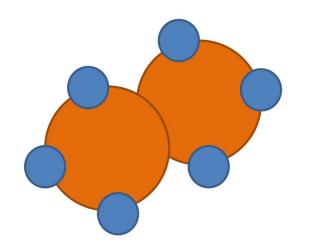
How many atoms?

How many elements?

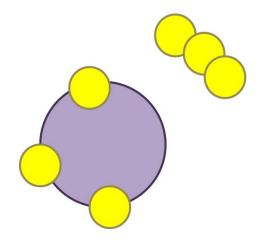


How many atoms?

How many elements? 2

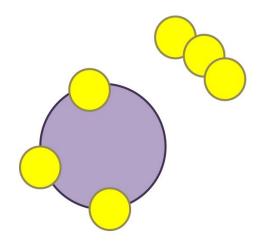


How many atoms?	
How many elements?	
How many molecules?	



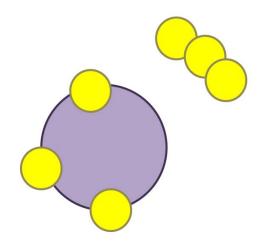
How many atoms?

How many elements?



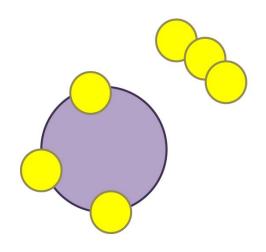
How many atoms?

How many elements?

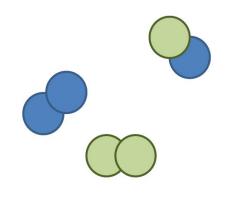


How many atoms?

How many elements?

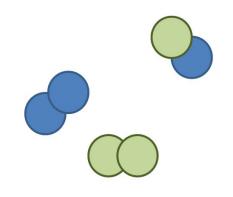


How many atoms?	
How many elements?	
How many molecules?	



How many atoms?

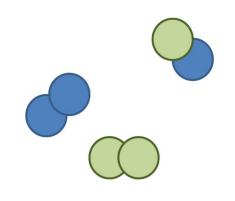
How many elements?



How many atoms?

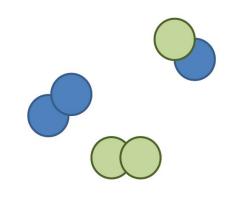
How many elements?

How many molecules?



How many atoms?

How many elements?



How many	atoms?	
How many	elements?	
How many	molecules?	

Now, what about compounds?

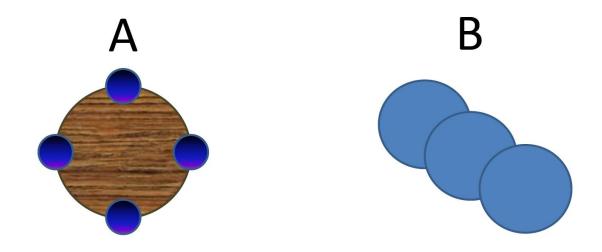
# Now, what about compounds?

Compounds are any group of atoms (or molecule) that has more than one type of atom.

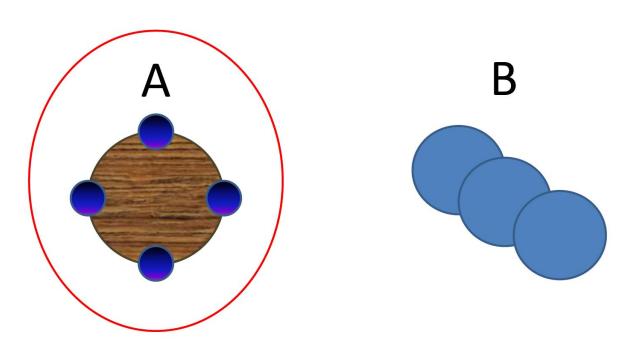
#### Understanding Elements and Compounds



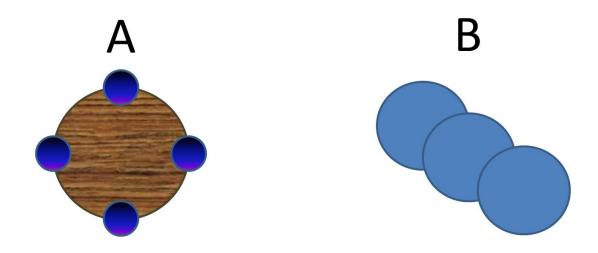
### Which one is a compound?



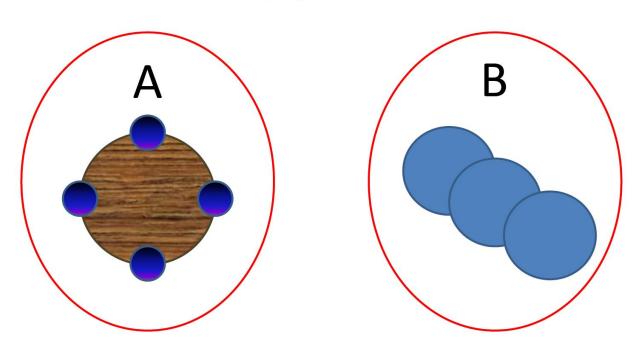
# Which one is a compound?

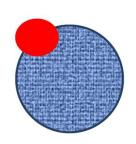


#### Which one is a molecule?



#### Which one is a molecule?

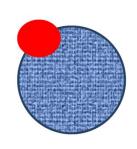




How many atoms?

How many elements?

How many molecules?

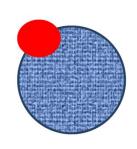


How many atoms?

3

How many elements?

How many molecules?

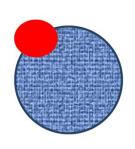




How many atoms?

How many elements?

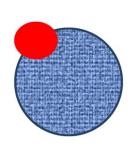
How many molecules?



How many atoms?

How many elements?

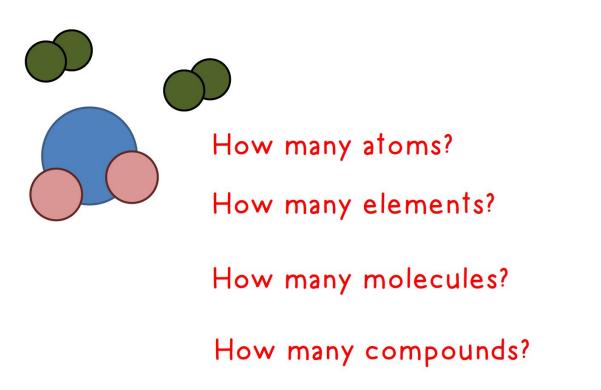
How many molecules?

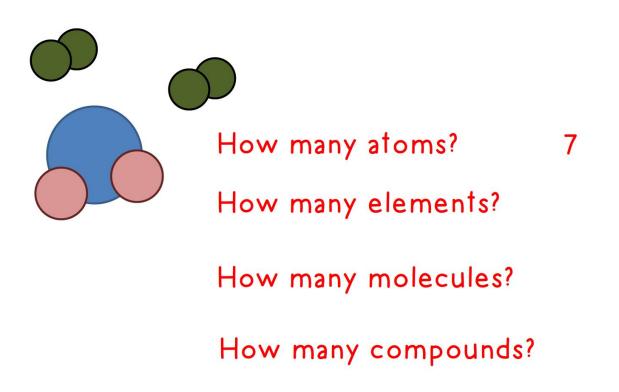


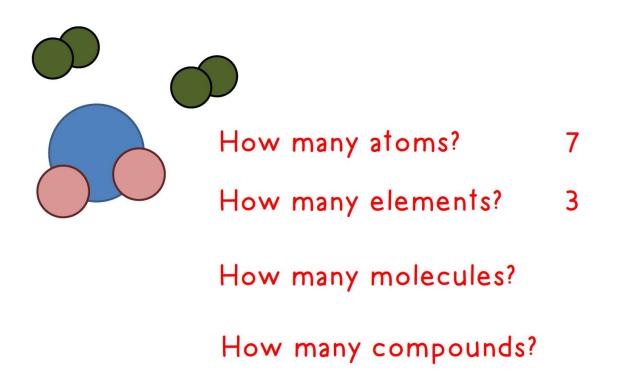
How many atoms?

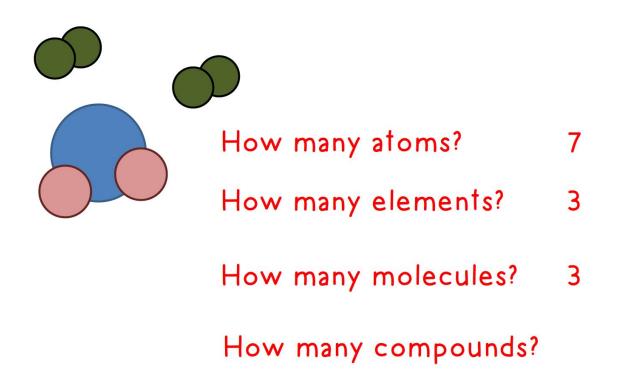
How many elements?

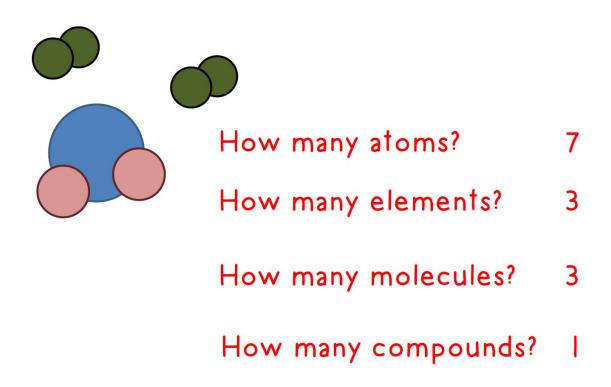
How many molecules?

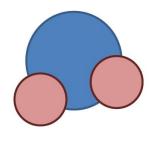


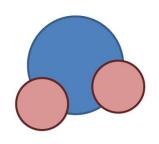








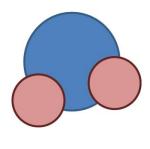


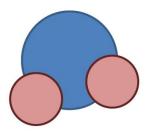


How many atoms?

How many elements?

How many molecules?

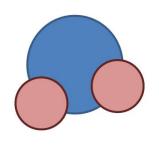


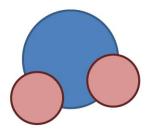


How many atoms?

How many elements?

How many molecules?

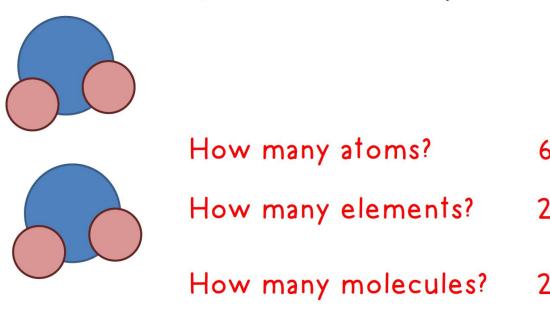


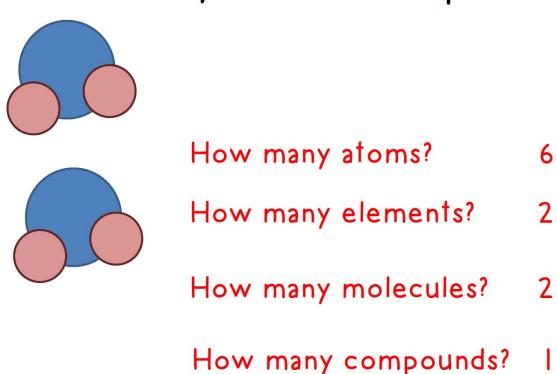


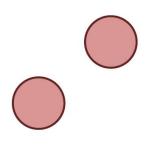
How many atoms? 6

How many elements?

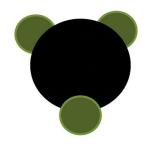
How many molecules?





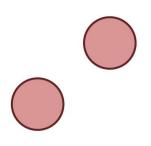


How many atoms?

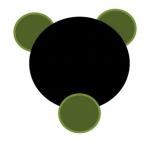


How many elements?

How many molecules?

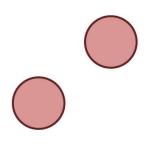


How many atoms?

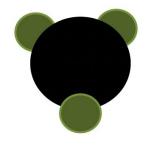


How many elements?

How many molecules?

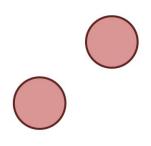


How many atoms?

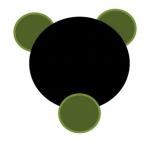


How many elements?

How many molecules?

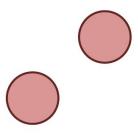


How many atoms?

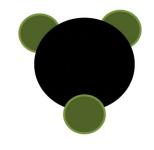


How many elements? 3

How many molecules?







How many elements? 3

How many molecules?

#### What is a "subscript"?



The subscript '2' tells us that there are two H atoms in one molecule of water

O has no subscript; that means there is just one O atom in a molecule of water Need extra help before starting today's activity?

Scan this QR code



Grab some headphones & watch this video!